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APPLICATION NO.	F	ILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO	
09/674,547	01/24/2001		Masaru Kawai	30681-1005	30681-1005 8985	
5179	7590	01/14/2004		EXAMINER		
PEACOCK MYERS AND ADAMS P C P O BOX 26927				HAMLIN, DERRICK G		
ALBUQUERQUE, NM 871256927				ART UNIT	PAPER NUMBER	
				1751	1751	

DATE MAILED: 01/14/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)					
Office Action Summary	09/674,547	KAWAI ET AL.					
Office Action Summary	Examiner	Art Unit					
The SEAU INC DATE of this communication can	Derrick G. Hamlin	1751					
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the t	correspondence address					
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION.  Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication.  If the period for reply specified above is less than thirty (30) days, a reply If NO period for reply is specified above, the maximum statutory period Fallure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).  Status	66(a). In no event, however, may a reply be till within the statutory minimum of thirty (30) day ill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	mely filed rs will be considered timely. the mailing date of this communication. DD (35 U.S.C. § 133).					
1) Responsive to communication(s) filed on <u>05 L</u>	December 2003 .						
2a) ☐ This action is <b>FINAL</b> . 2b) ☑ Thi	s action is non-final.						
Since this application is in condition for alloward closed in accordance with the practice under a secondary.							
Disposition of Claims							
4) Claim(s) <u>1-31</u> is/are pending in the application							
	4a) Of the above claim(s) is/are withdrawn from consideration.						
· - · · · · · · · · · · · · · · · · · ·	Claim(s) is/are allowed.						
<u> </u>	Claim(s) <u>1-31</u> is/are rejected.						
7) Claim(s) is/are objected to.							
<ul> <li>8) Claim(s) are subject to restriction and/or Application Papers</li> </ul>	election requirement.						
9) The specification is objected to by the Examiner							
10) The drawing(s) filed on is/are: a) accept		miner					
, , , , , , , , , , , , , , , , , , , ,							
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  11) The proposed drawing correction filed on is: a) approved b) disapproved by the Examiner.							
If approved, corrected drawings are required in rep		- · <b>- ,</b> · · · · · · · · · · · · · · · · ·					
12) The oath or declaration is objected to by the Exa	aminer.						
Priority under 35 U.S.C. §§ 119 and 120							
13) Acknowledgment is made of a claim for foreign	priority under 35 U.S.C. § 119(a	a)-(d) or (f).					
a) ☐ All b) ☐ Some * c) ☐ None of:	, , , , , , , , , , , , , , , , , , , ,	, , , , ,					
1. Certified copies of the priority documents	have been received.						
2. Certified copies of the priority documents	2. Certified copies of the priority documents have been received in Application No						
Copies of the certified copies of the prior application from the International Bur     See the attached detailed Office action for a list of the certified in the certified copies of the certified copies of the prior applied in the certified copies of the ce	eau (PCT Rule 17.2(a)).						
14) ☐ Acknowledgment is made of a claim for domestic	priority under 35 U.S.C. § 119(	e) (to a provisional application).					
a) The translation of the foreign language pro	• • •						
Attachment(s)	. 30						
1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO-1449) Paper No(s)	5) Notice of Informal	y (PTO-413) Paper No(s) Patent Application (PTO-152)					
S. Patent and Trademark Office	Section Section 1 11						

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#### **DETAILED ACTION**

#### Request for Continued Examination

The request filed on 12/5/2003 for a Request Continued for Continued

Examination (RCE) under 37 CFR 1.114 based on parent Application No. 09/674,547 is

acceptable and a RCE has been established. An action on the RCE follows.

### Claim Rejections - 35 USC § 112

Claim 1 is rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention. It is not clear if the applicant is referring to the percent water or percent ethylene glycol. As written it appears to redundantly claim that the water is present between 15-50% by weight of ethylene glycol. It also appears as if the range may apply to both the ethylene and propylene glycol. The applicant may have inserted the new limitation in the wrong place and clarification is required.

## Claim Rejections - 35 USC § 103

The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

The rejection of claims 1-31 under 35 U.S.C. 103(a) as being unpatentable over Burns et al. (5085793), is maintained for the reasons set forth in the rejection filed

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10/02/2002. The following rejection will be maintained until claim 1 is clarified it reflect the amounts of each component that the applicant wants to claim.

The applicant argues that automobile coolant is added to an engine concentrated and then water is added separately and the running engine mixes the chemicals. The applicant argues that his composition is diluted already and contains 50% deionized water. However, Burns teaches an embodiment that is diluted with 10-90% water (col. 5, lines 5-9) and all the examples in Table I require deionized water.

Additionally, the applicant is not claiming a method of making nor a process for mixing, therefore the final composition in the car would be the claimed composition.

Again, the applicant has not furnished a declaration from the inventor or one of ordinary skill in the art, to show demonstrating that the composition in the engine would not be the composition that is instantly claimed.

Burns teaches a corrosion-inhibited antifreeze composition. Specifically, the antifreeze composition comprises a major portion of a liquid alcohol freezing point depressant and a minor portion of at least one hydroxyl-substituted aromatic carboxylic acid having the hydroxyl radical disposed proximate to the carboxylic radical. The above described acid is employed as a corrosion inhibitor. (abstract) The reference also discloses that the freezing point depressant is selected from the group consisting of ethylene glycol, diethylene glycol, propylene glycol, etc. and that the composition way be 80 to about 99 weight percent of a liquid alcohol freezing point depressant and from about 20 to about 1 weight percent of a corrosion inhibitor, said corrosion inhibitor consisting of (a) at least one hydroxyl-substituted aromatic carboxylic acid, and (b)

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member selected from the group consisting of alkali metal borates, alkali metal silicates, alkali metal benzoates, alkali metal nitrates, alkali metal nitrites, alkali metal molybdates, hydrocarbyl thiazoles and mixtures thereof, a C8 -C12 aliphatic dibasic acid or the alkali metal, ammonium or amine salt of said acid (col. 5, lines 24-28 and 48-59). The reference teaches several other conventional corrosion inhibitors may be employed in conjunction with the acids which are well known in art, such as alkylbenzoic acid or the alkali metal, ammonium or amine salt thereof; C8 to C12 aliphatic monobasic acid or the alkali metal, ammonium or amine salt thereof and a hydrocarbyl triazole and; alkali metal salt of benzoic acid, an alkali metal salt of a dicarboxylic acid and an alkali metal nitrate; alkali metal carbonates, borax, the alkali metal dichromates, the alkali metal silicates, phosphorus acid, phosphoric acid; an alkali metal tungstate, benzotriazole, tolyltriazole, an alkali metal salt of benzoic or toluic acid, an alkali metal salt of a phenol, an alkanolamine and an organo-silicone compound; and 4-tert-butylbenzoic acid (col. 1, line 31 – col. 2, lines 55).

The reference fails to teach the specific use of cinnamic acids. The reference does not teach which corrosion inhibitors must be included or excluded. The reference also fails to teach the instantly claimed amounts.

Although the reference fails to teach the specific use of cinnamic acids they are encompassed in the general teaching of alkylbenzoic acid. The reference does not teach which corrosion inhibitors must be included or excluded, nor does it require anything more than a one hydroxyl-substituted aromatic carboxylic acid. The reference also fails to teach the instantly claimed amounts for each specific component, however

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the reference does teach the minimum and maximum amounts of liquid alcohol freezing point depressant corrosion inhibitor and instantly claimed amounts clearly overlap with the reference

Therefore it would have been obvious to one of ordinary skill in the art at the time the invention was made to make the instantly claimed prediluted solutions, since the reference teaches a corrosion-inhibited antifreeze composition which may contain a glycol freezing point depressant and a minor portion of at least one hydroxyl-substituted aromatic carboxylic acid and additional corrosion inhibitors.

In view of the forgoing, the above claims have failed to be patently distinguishable over prior art.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Derrick G. Hamlin whose telephone number is (571) 272-1317. The examiner can normally be reached on Monday-Thursday and alternating Fridays from 8:30 AM - 5:00 PM.

If reasonable attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Dr. Yogendra Gupta, can be reached on (571) 272-1316. The fax phone number for this Group is (703) 872-9310.

Any inquiry of a general nature or relating to the status of this application should be directed to the Group receptionist whose telephone number is (703) 308-0661.

Derrick G. Hamlin

1/11/04

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